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Three of us have had the pleasure of working with the folks at W. H. Freeman and Company on a number of projects, whereas one of us is new to the Freeman family. Our experiences have always been delightful and rewarding. Writing and producing the seventh edition of *Biochemistry* was no exception. The Freeman team has a knack for undertaking stressful, but exhilarating, projects and reducing the stress without reducing the exhilaration and a remarkable ability to coax without ever nagging. We have many people to thank for this experience. First, we would like to acknowledge the encouragement, patience, excellent advice, and good humor of Kate Ahr Parker, Publisher. Her enthusiasm is source of energy for all of us. Lisa Samols is our wonderful developmental editor. Her insight, patience, and understanding contributed immensely to the success of this project. Beth Howe and Erica Champion assisted Lisa by developing several chapters, and we are grateful to them for their help. Georgia Lee Hadler, Senior Project Editor, managed the flow of the entire project, from copyediting through bound book, with her usual admirable efficiency. Patricia Zimmerman and Nancy Brooks, our manuscript editors, enhanced the literary consistency and clarity of the text. Vicki Tomaselli, Design Manager, produced a design and layout that makes the book exciting and eye-catching while maintaining the link to past editions. Photo Editor Christine Beuse and Photo Researcher Jacalyn Wong found the photographs that we hope make the text more inviting. Janice Donnola, Illustration

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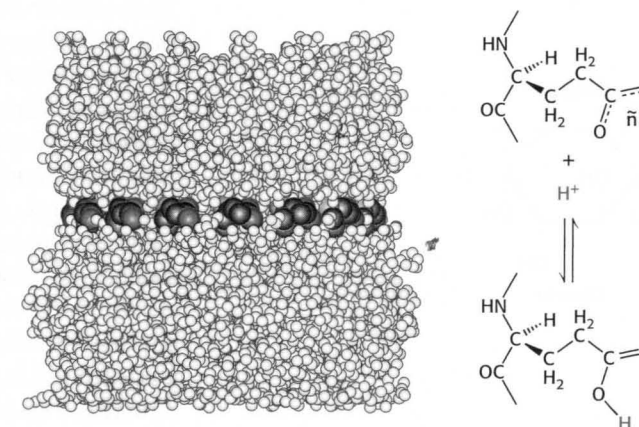
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Biochemistry: An Evolving Science



Chemistry in action. Human activities require energy. The interconversion of different forms of energy requires large biochemical machines comprising many thousands of atoms such as the complex shown above. Yet, the functions of these elaborate assemblies depend on simple chemical processes such as the protonation and deprotonation of the carboxylic acid groups shown on the right. The photograph is of Nobel Prize winners Peter Agre, M.D., and Carol Greider, Ph.D., who used biochemical techniques to study the structure and function of proteins. [Courtesy of Johns Hopkins Medicine.]

Biochemistry is the study of the chemistry of life processes. Since the discovery that biological molecules such as urea could be synthesized from nonliving components in 1828, scientists have explored the chemistry of life with great intensity. Through these investigations, many of the most fundamental mysteries of how living things function at a biochemical level have now been solved. However, much remains to be investigated. As is often the case, each discovery raises at least as many new questions as it answers. Furthermore, we are now in an age of unprecedented opportunity for the application of our tremendous knowledge of biochemistry to problems in medicine, dentistry, agriculture, forensics, anthropology, environmental sciences, and many other fields. We begin our journey into biochemistry with one of the most startling discoveries of the past century: namely, the great unity of all living things at the biochemical level.

OUTLINE

- 1.1 Biochemical Unity Underlies Biological Diversity**
- 1.2 DNA Illustrates the Interplay Between Form and Function**
- 1.3 Concepts from Chemistry Explain the Properties of Biological Molecules**
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1.1 Biochemical Unity Underlies Biological Diversity

The biological world is magnificently diverse. The animal kingdom is rich with species ranging from nearly microscopic insects to elephants and whales. The plant kingdom includes species as small and relatively simple